

Electronic Properties of Carbon at Extreme Conditions from ab Initio Simulations

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Electronic properties of Carbon at extreme conditions from ab initio simulations

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APS March meeting, Montreal 2004



Motivation

Why Carbon at extreme conditions?

- Astrophysics and Planetary Science: Jovian Planets¹ and White Dwarfs²
- **Technology:** High-pressure research, transparency problem³

- Practical: Experiments are difficult at extreme conditions and
- Methodology: Empirical/non-abinitio simulations gave misleading



¹W. B. Hubbard, *Science* **214**, 145-9 (1981)

²V. Weidemann et al., Astron. & Astroph. 297(1), 216-222 (1995)

³A. L. Ruoff et al., *J. Appl. Phys.* **70**, 2066 (1991)

⁴C. J. Wu et al., PRL **89**, 135701 (2002) Correa, Boney, Galli, Falcone

Motivation

Why Carbon at extreme conditions?

- Astrophysics and Planetary Science: Jovian Planets¹ and White Dwarfs²
- Technology: High-pressure research, transparency problem³

Why ab initio simulation?

- Practical: Experiments are difficult at extreme conditions and results are controversial.
- Methodology: Empirical/non-abinitio simulations gave misleading results⁴.

Electronic Gap dependence with Pressure and Temperature



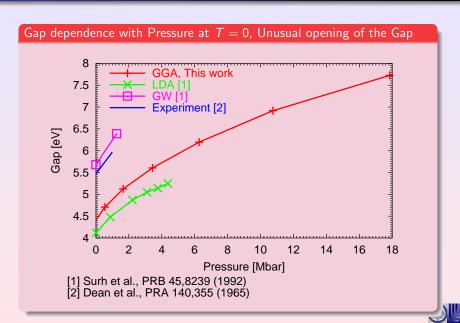
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Gap of Diamond increases with Pressure



1. sp^3 hybridization

- Simple Bonding-Antibonding picture works for diamond⁵.
- Silicon shows opposite behaviour.

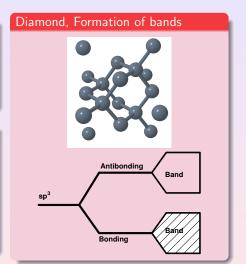
- Under Anisotropic pressure the gap decreases.
- BC8 shows a decrease of the gap.
- Thermal disorder produces a rapid decrease of the gap



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2. Symmetry of Diamond structure

- Under Anisotropic pressure the gap decreases.
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BC8 structure, stable above 12Mbar 3GA Gap [eV]



20

Pressure [Mbar]

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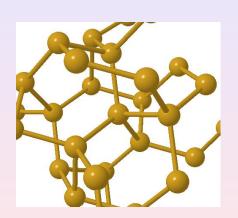
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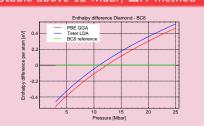
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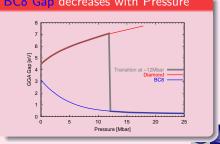
BC8 Structure, 4-fold coord. but different to Diamond



Stable above 12 Mbar, ΔH method



BC8 Gap decreases with Pressure



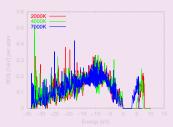
Gap of Diamond at Finite Temperature: Method

Constant Pressure Molecular Dynamics

- 64 atoms in a periodic box
- Car-Parinello molecular dynamics
- Norm conserving pseudo-potential
- Set $P = \sim 10 \text{Mbar}$ and T = 0 up to 10000 K
- ~4 picoseconds **equilibration** time



- DOS calculated from Snapshots
- GGA Energy Gap from DOS.





Gap of Diamond at Finite Temperature: Method

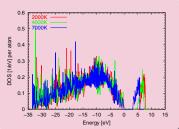
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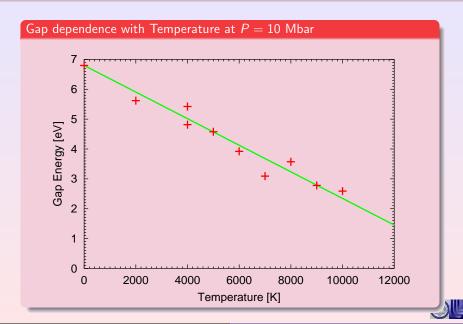
Density of States

- DOS calculated from Snapshots
- GGA Energy Gap from DOS.





Gap of Diamond reduces with Temperature at High P

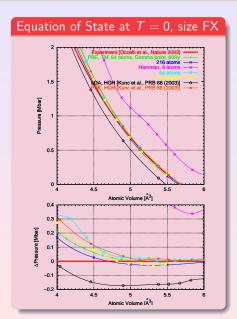


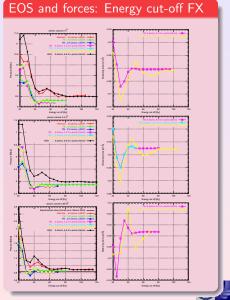
Summary

- Energy Gap of Carbon is very sensitive to crystal symmetry.
- BC8 Gap decreases with pressure.
- Diamond Gap increases with pressure.
- but decreases continuously as temperature is raised.
- Diamond at High pressure does not become conducting before melting



Appendix: Convergence tests





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